

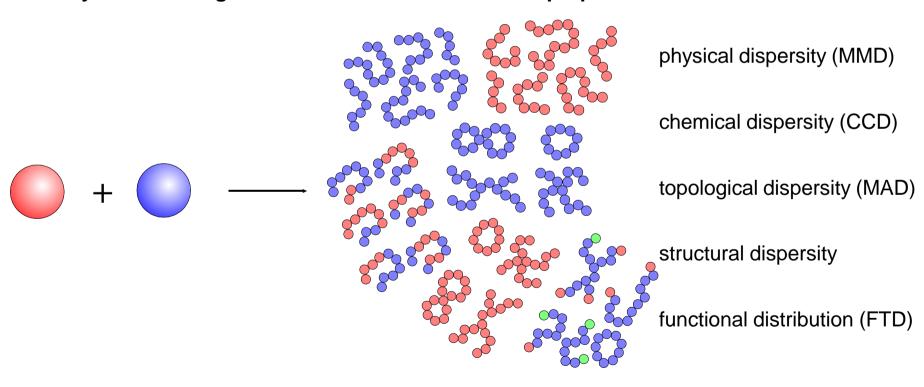
Size-exclusion Chromatography as a Useful Tool for the Assessment of Polymer Quality and Determination of Macromolecular Properties

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- 1. Introduction into Macromolecular Chromatography
- 2. Separation and 2-dimensional Techniques
- 3. Detection and Information Content
- 4. Summary



Most polymeric materials are highly complex multi-component materials even simple polymerization leads to products with multiple property distributions Analytical Challenges: Determination of distributed properties





Characterization Strategy

A) Characterization of bulk materials

\dc

access to bulk properties / property averages

requires batch methods

e.g. Light Scattering (LS), Viscometry, Osmometry, Ultracentrifugation (AUC)

NMR, IR, ...

B) Characterization of separated fractions

requires comprehensive chromatography

1) analytical fractionation methods:

access to property distributions

E.g. Liquid chromatography (SEC, LAC, LCCC)

Ultracentrifugation (AUC) Field flow fractionation (FFF) (Gas chromatography: GC)

Mass sprectrometry: MALDI-ToF

2) detection techniques

e.g. RI, UV, LS, Viscometry, FTIR, NMR, MS

separation - detection combinations determine which distributions can be measured



Chromatographic Modes

a) Size exclusion mode: SEC

$$K_{SEC} = exp(\Delta S/R)$$

$$0 < K_{SEC} < 1$$
 $\Delta H = 0$

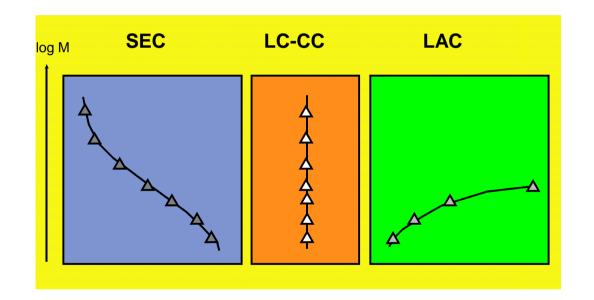
b) Adsorption mode: HPLC

$$K_{HPLC} = exp (- \Delta H/RT)$$

$$K_{HPLC} > 1$$
 $\Delta H \gg T\Delta S$



$$K = 1$$
 $\Delta H = \Delta S$





Chromatographic Modes of Separation

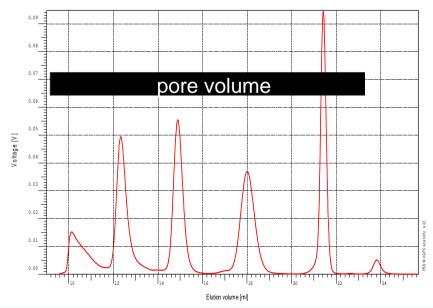
Comparison of Chromatographic Modes

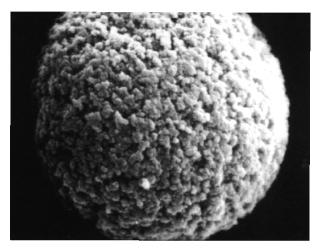
technique	separation governed by	information content	potential problems
SEC	 hydrodynamic volume molecular size in solution diffusion controlled process 	molar mass (MMD)chemical composition (CCD)	calibration dilemmaspecific interactions
LC-CC	 chain inhomogeneity defect structures endgroups diffusion and adsorption controlled process 	functionality type (FTD)molecular architecture (MAD)	 irreversible adsorption determination of critical adsorption point
HPLC	chemical compositionendgroup	chemical composition (CCD)functionality type (FTD)	 molar mass influence partial adsorption
	adsorption controlled process	S	• large k'

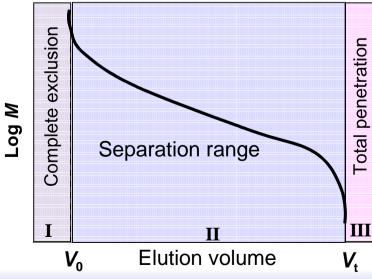


SEC Separation Principles

- solutes diffuse between mobile phase and pores in stationary phase
- conformational entropy loss is driving force
- retention based on hydrodynamic size in solution V_h
- molar mass by rentention calibration or proper detection method









SEC Instrumentation

Special instrumental requirements:

- solvent compatibility
- prevent clogging by solvent evaporation
- multi-detector application
- columns: mainly polymer packing
- often: absolute concentrations required absolute injection volume required

critical modules:

- pumps
- autosamplers
- software

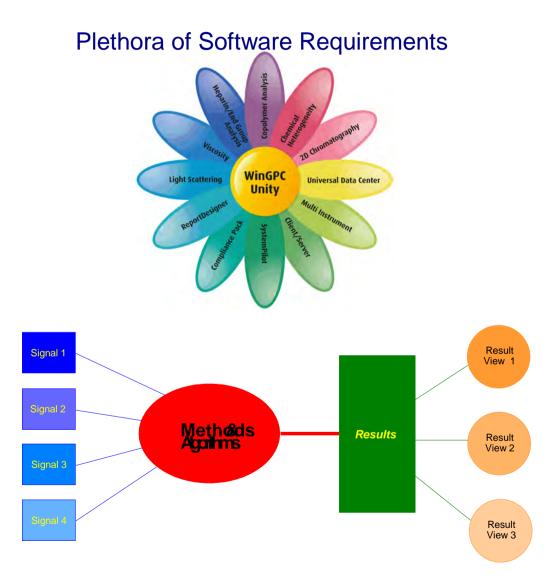




SEC Data Systems

Special software requirements:

- long analysis times
- complex data treatment
- multi-signal processing
- determination of distributions
- combination of methods
- multiple vendor support
- integration in existing infrastructure





Determination of Property Distributions

complete description of properties and contributions

- accurate determination of amounts
- proper measurement/calibration of properties
- accurate results calculation and representation

Example: Conversion of raw signals to molar mass distribution

The $molecular\ weight\ averages$ can be calculated from the moments, $\mu_i,$ of the molar mass distribution:

$$\mu_i = \int_0^\infty M^i \cdot w(M) \, \mathrm{d} M$$

with: µ, the i-th moment of the molar mass distribution

The molar mass averages are defined and calculated in PSS WinGPC Unity by: Number average molecular weight:

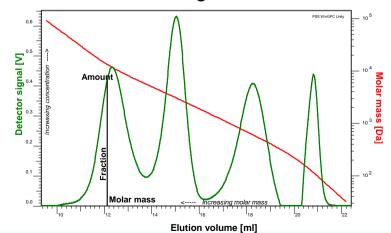
$$M_{\rm M} = \frac{\sum h(M) \cdot M}{\sum h(M)} = \frac{\sum w(M)}{\sum w(M) / M} = \frac{\mu_0}{\mu_{-1}}$$

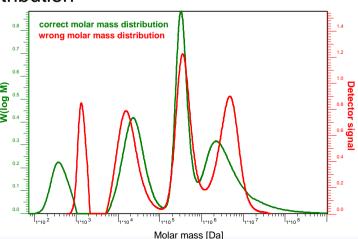
Weight average molecular weight:

$$M_{\mathbf{w}} = \frac{\sum h(M) \cdot M^2}{\sum h(M) \cdot M} = \frac{\sum w(M) \cdot M}{\sum w(M)} = \frac{\mu_1}{\mu_0}$$

z-average molecular weight:

$$M_x = \frac{\sum h(M) \cdot M^3}{\sum h(M) \cdot M^2} = \frac{\sum w(M) \cdot M^2}{\sum w(M) \cdot M} = \frac{\mu_2}{\mu_1}$$



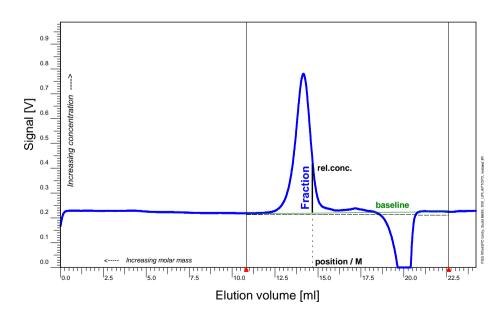


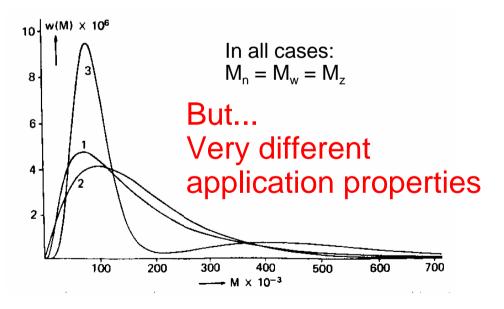


Determination of fundamental parameters

Chromatogram: calibration curve: molar mass distribution:

relates apparant concentration to elution volume / retention time relates molar mass to chromatographic position shows mass fraction of molecules of given molar mass





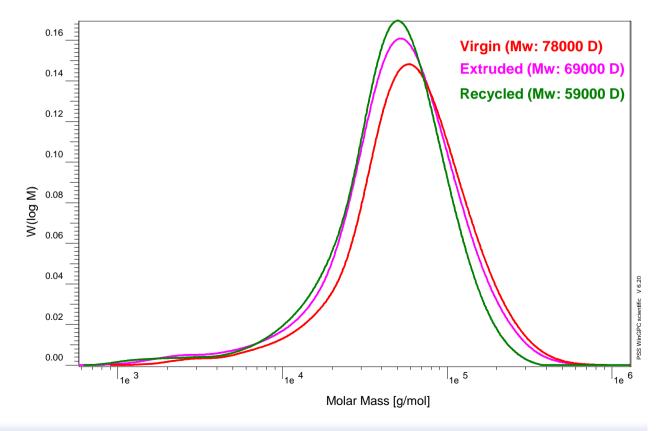


Polymer Degradation during Recycling Processes

conditions:

system: PSS SECcurity GPC

eluent: TCM/HFIP
columns: PSS SDV 5µm
detection: UV@260nm
software: PSS WinGPC
analysis in: 35 min / sample



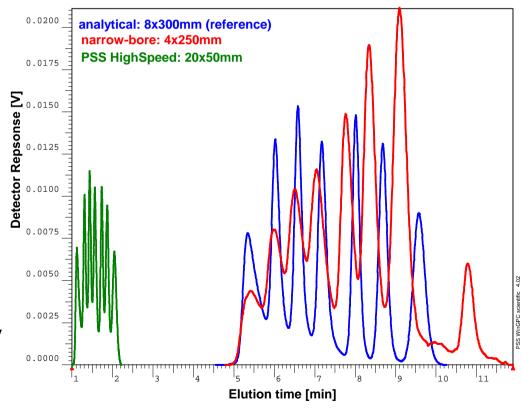


Conventional and HighSpeed Analysis

Benefits of HighSpeed SEC

- short analysis time (up to 10-fold)
- no (additional) sample degradation*
- no special SEC hardware required*
- no method change*

*) only by using PSS HighSpeed column technology





Quality Assurance by HighSpeed SEC

commercial polycarbonate in THF mw by producer: 30000 g/mol

60 repeats in 2h

column: 2x PSS SDV 5 µm HighSpeed

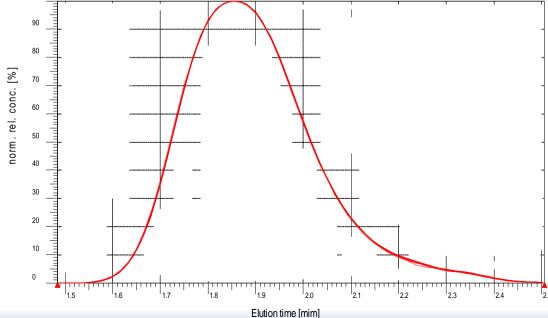
calibration: PSS ReadyCal PS standards

detection: UV

HighSpeed result:

M_w: (29610±150) g/mol

RSD: 0.5%



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www.pss-polymer.com



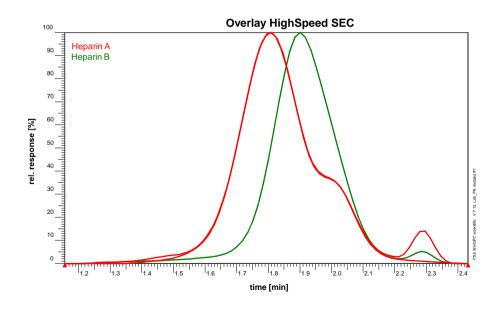
HighSpeed Heparin Quality Assurance

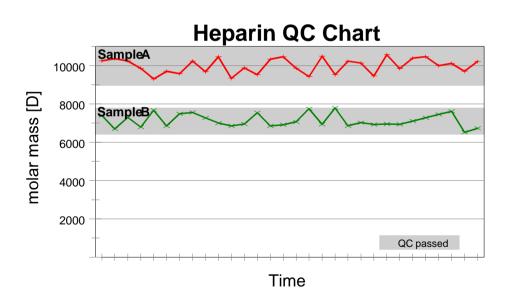
column: PSS HighSpeed Suprema 100, 10 µm

analysis time: 2 min

calibration: Heparin endgroup (DAB); PSS WinGPC

detection: RI







Chromatographic Modes of Separation

Potential limitations in SEC

- Molecular weight range:
 Separation range may be increased by using combination of different pore size columns
- Easy to overcome

- Resolution:
 - Separation efficiency may be increased by using longer or more columns of same pore size

Easy to overcome

$$n = 1 + \frac{\sqrt{L}}{4} \cdot \ln \frac{V_p}{V_0}$$

Size-separation
 co-elution of different species (e.g. copolymers, branched molecules) ifferent
 LC techniques



n independent properties require *n*-dimensional methods for accurate (independent) characterization.

Possible multidimensional chromatography techniques:

HPLC, SEC, LC-CC, GC, TREF, GPEC,.....

Example:

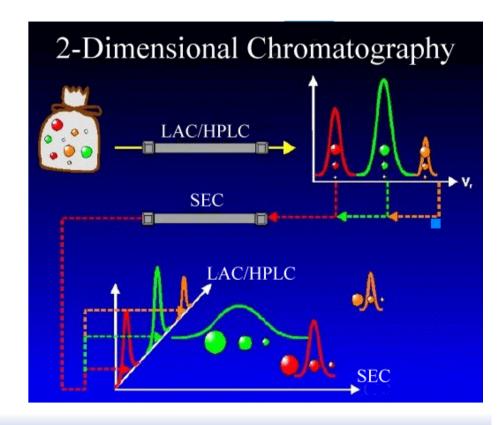
combination of LAC(HPLC) and SEC:

1st dimesion:

LAC/HPLC for separation according to CC

2nd dimesion:

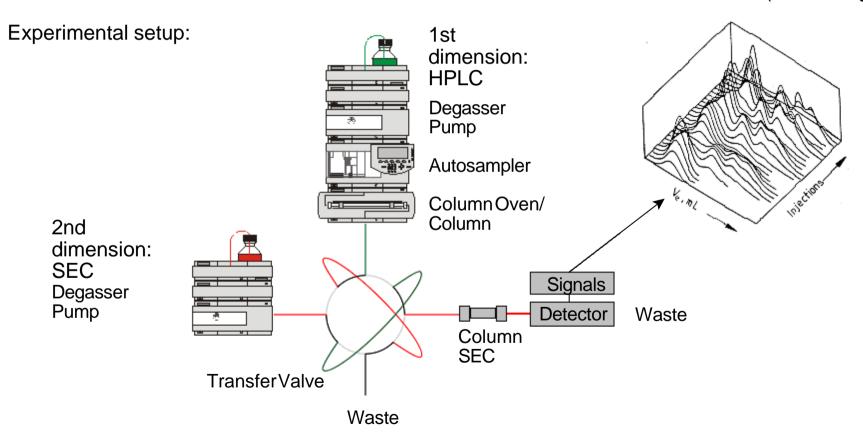
SEC for separation according to MM





Investigation of CCD and MMD

SEC results (chromatograms)





2D Chromatography

SEC Analysis of TPE

sampleBfailedinthefield

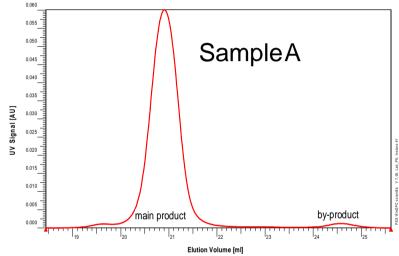
- main product looks very similar
- -similarby-productspresent

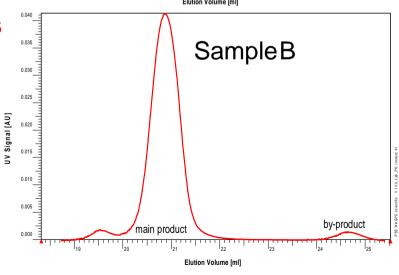
SEC does not track performance differences

sample A	sample B
00	00

Mn[kD] 99 90					
Mw[kD] 109 103					
Mw/Mn 1.08 1.14					
Mp[kD] 108 104					
by-product 0.8% 1.7%					
molar masses by narrow PSt calibration					

differences due to composition?







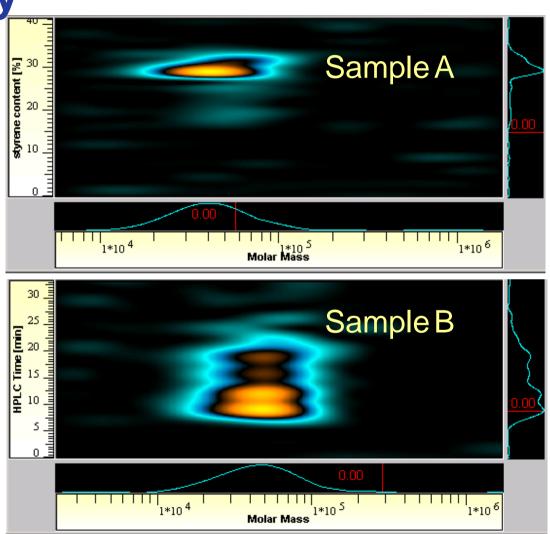
2D Chromatography

Comprehensive 2D by HPLCxSEC

HPLC tracking composition SEC tracking molar mass

2D analysis

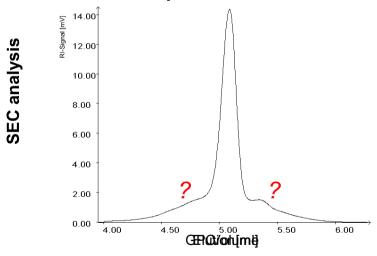
- corroborates similar MMD
- shows similar average PS content
- reveals big differences in CCD
- contour map shows
 - -differences easily
 - 2D property distributions



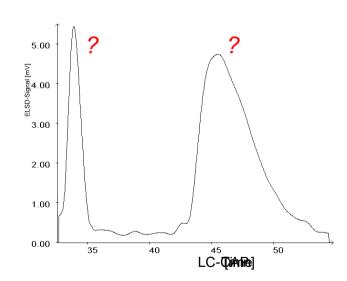


Investigation of by-product in motoroil additives









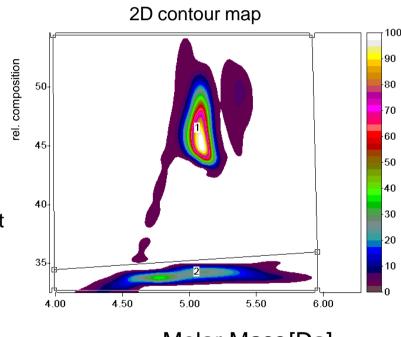
observations difficult to explain



Investigation of by-product in motor oil additives

2D results

- main product (region1)
- parallel reaction forms region 2
- two different processes
- by-product is homopolymer
- by-product has broad MMD
- reaction mixture contains 60% desired product
- desired product is narrow in CCD and MMD



Molar Mass[Da]



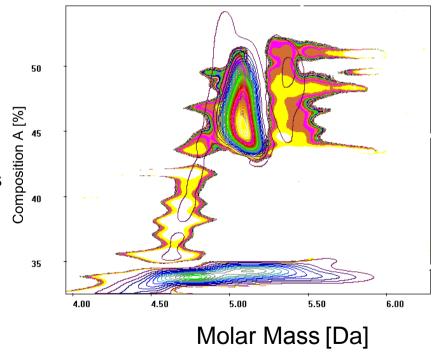
Investigation of by-product in motor oil additives

2D compositional analysis

overlay of 2D separation with chemical composition

supports

- two simultaneous polymerization processes
- desired product is copolymer
- by-product is homopolymer







Detector Signal Characteristics

$$U_d = K_d \times \sum_i (k_{Sample} \times c_{Sample} \times M^x)$$

U_d: Signal intensity

K_d: Instrument constant

k_{sample}: Sample dependent parameter

for spectroscopic detectors: $k_{sample} = extinction coefficient, \kappa$

for refractive index (RI) detectors: $k_{\text{sample}}^{\text{sample}} = \text{refractive index increment, dn/dc}$

note: dependent on solvent composition, Τ, λ

c_{sample}: Sample concentration

M: Molar mass

x: Detector dependent

for RI, UV, ELSD: X = 0for on-line LS and MS detectors: X = 1

for on-line viscometers: $X = Mark Houwink coefficient \alpha$

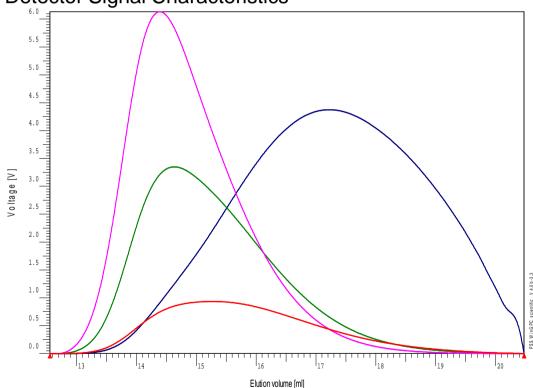
for on-line NMR, osmometers* : X = -1

* not commercially available



Detector Properties

Detector Signal Characteristics



concentration detector

Refractive index detector (RI)

molar mass sensitive detectors
On-line light scattering detector

On-line viscosimeter

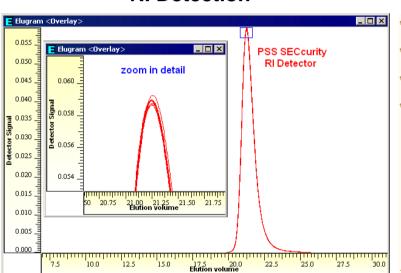
On-line mass spec (or osmometer)

advanced detector combinations provide comprehensive molecular and structural information

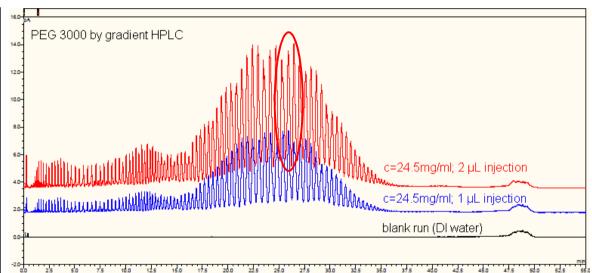


Requirements for Accurate Quantification

RI Detection



Corona (ELS) Detection



Linear response (conc-area)
Stable signal (high repeatability)
No molar mass influence
non-specific detector

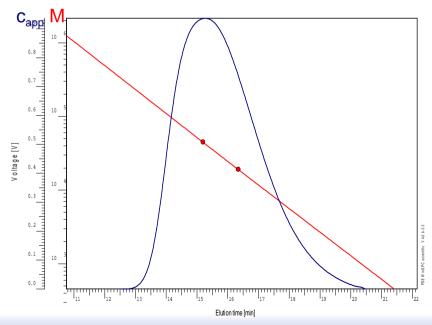
Strong non-linear response (even log-log)
Poor signal stability (low repeatability)
Molar mass dependent
only non-volatiles detected



Multiple Detection in SEC Mode

 $c(V), M_c(V) \rightarrow x_k(M), w(log M_c), M_{n,c}, M_{w,c}, D_c$ $c_{app}(V), M(V)$ What we need:

What we have:



advantages:

- uses ordinary SEC equipment
- copolymer analysis with same injection
- no additional sample preparation

limitations:

- statistical copolymers
- graft copolymers with high graft density



Multiple Detection in SEC Mode

Approach:

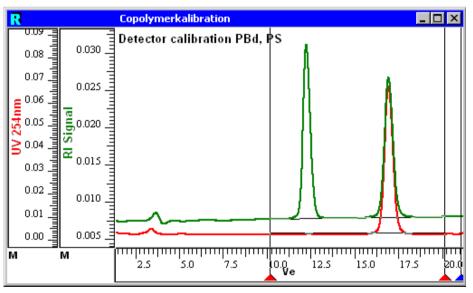
Task1: derive true c(V) from $c_{app}(V)$ needs multi-detector setup with detector calibration

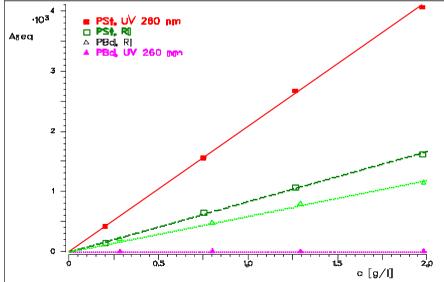
 \rightarrow absolute concentration of all components k in sample



Multiple Detection in SEC Mode

Determination of copolymer response factors







Multiple Detection in SEC Mode

Determination of comonomer concentrations

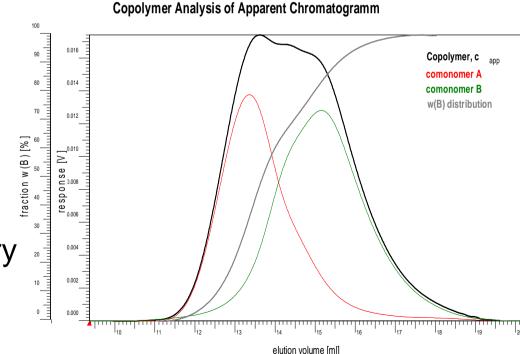
$$c_{app} = \sum_{k} f_{dk} \cdot c_{k}$$

advantages:

- universal approach
- no special equipment necessary



neighbor-group effects



IUPAC Workshop, MACRO 2016, Istanbul

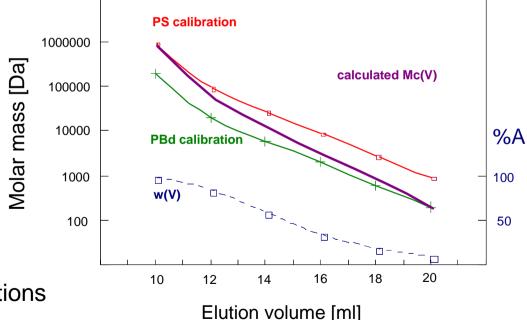
www.pss-polymer.com



Multiple Detection in SEC Mode

Task 2: $M_c(V)$ from homo polymer calibration, or $M_c(V)$ directly from molar mass sensitive detection

$$Ig M_c(V) = \sum w_k(V) \cdot M_k(V)$$



correct for negligible hetero-contact interactions



DeterminatiomChemicaHeterogeneity

Investigation of ABA block copolymer in SEC Mode

SEC results with PS standards:

Mn 127 kDa

Mw 353 kDa

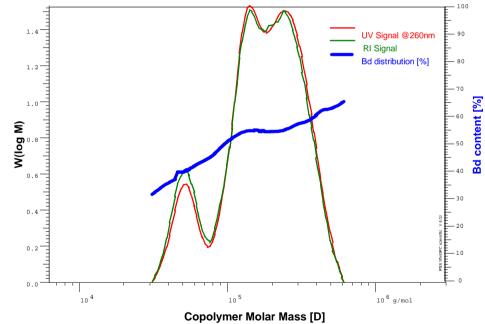
PD 2.78

Copolymerresults with multidetection:

Mn 76.3 kDa

Mw 222 kDa

PD 2.91



by PS and PBd calibration



SEC with a light scattering detector: MMD, MAD information

Theoretical Background Light Scattering:

for monodisperse samples, diluted solutions, particle size $< \lambda/20$

$$R(\theta) = K \cdot c \cdot M$$

Optical constant, includes refractive index increment (dn/dc)²

M: Molar Mass

Concentration

for polydisperse samples with larger partical size (non-isotropic scatterer):

$$K \cdot c/R(\theta) = 1/M_w [1+16/3 \pi^2/\lambda^2 < R^2 >_z sin^2(\theta/2)] + 2 A_2 \cdot c$$



SEC with a light scattering detector: MMD, MAD information

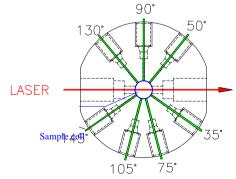
Theoretical Background Light Scattering:

$$K \cdot c/R(\theta) = 1/M_w [1+16/3 \pi^2/\lambda^2 < R^2 >_z sin^2(\theta/2)] + 2 A_2 \cdot c$$

Light scattering techniques:

LALLS: Low angle laser light scattering RALLS: Right angle laser light scattering MALLS: Multi angle laser light scattering

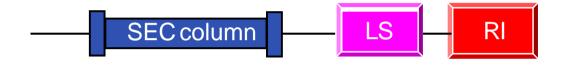
MALLS: PSS SLD7000 detector cell





SEC with a light scattering detector:

MMD, MAD information

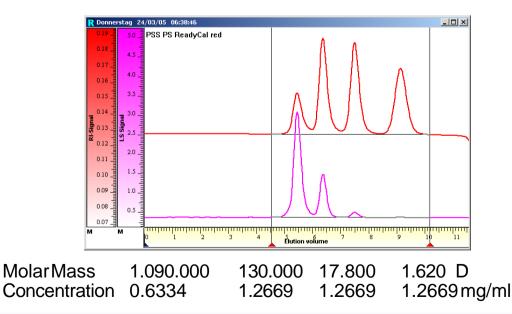


LS can be MALLS, RALLS, LALLS

LS signal:
$$U(LS) = K' \cdot (dn/dc)^2 \cdot c \cdot M$$

RI singal:
$$U(RI) = K'' \cdot c$$

$$\frac{LS - Signal}{RI - Signal} \rightarrow M \cdot (dn / dc)^{2}$$

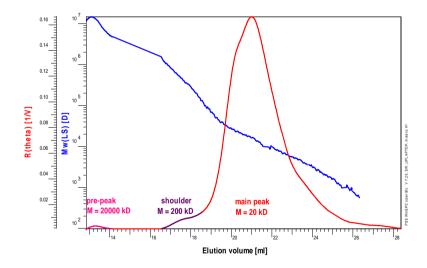




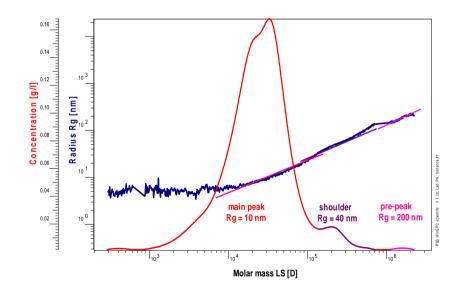
SEC with a light scattering detector: MMD, MAD information

PVB (Poly-vinyl butyral) sample: SEC-MALLS

Results on-line Zimm plot:



molar mass measured for every fraction MMD



radius of gyration measured for every fraction MAD



SEC with a viscometer detector: MMD, MAD information

Theoretical Background:

SEC separates according to hydrodynamic volume

$$V_{h,1} = V_{h,2}$$

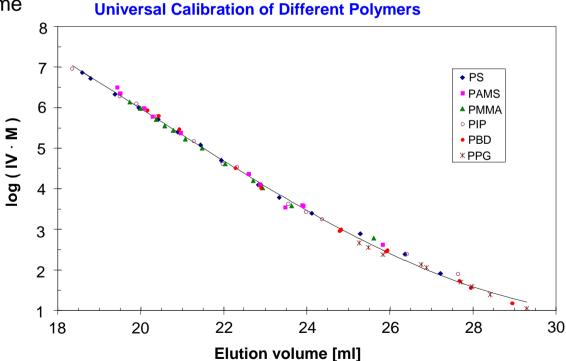
$$[\eta]_1 \cdot M_1 = [\eta]_2 \cdot M_2$$

A chance to solve the calibration dilemma: Universal calibration curve

$$M_2 = [\eta]_1 \cdot M_1 / [\eta]_2$$

$$[\eta]_2 = \mathsf{K} \cdot \mathsf{M}_2^{\alpha}$$

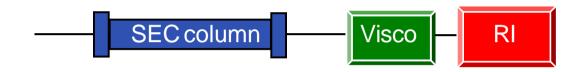
Mark-Houwink equation Structure information





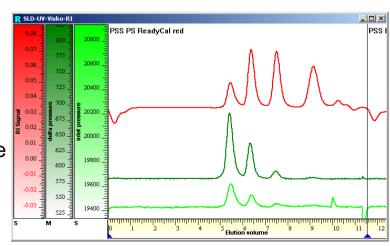
SEC with a viscometer detector:

MMD, MAD information



Viscometer signal: $U(V) = K' \cdot [\eta] \cdot c$

RI signal: $U(RI) = K'' \cdot c$





SEC with a viscometer detector: MMD, MAD information

Structure information, MAD:

 $[\eta] = K \cdot M^{\alpha}$ Mark-Houwink equation

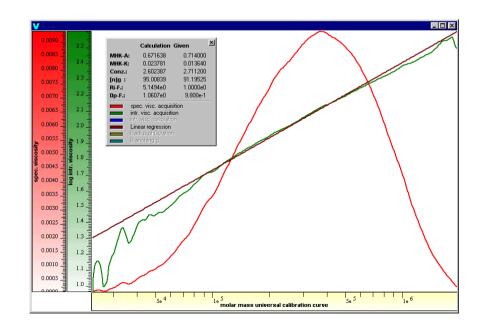
 $\alpha = 2$ rigid rod

 $1 > \alpha > 0.5$ random coil

 $\alpha = 0.5$ random coil, Theta-conditions

 $\alpha = 0$ solid sphere

Branching coefficient g': $g' = \{[\eta]_{branched}/[\eta]_{linear}\}_{M}$

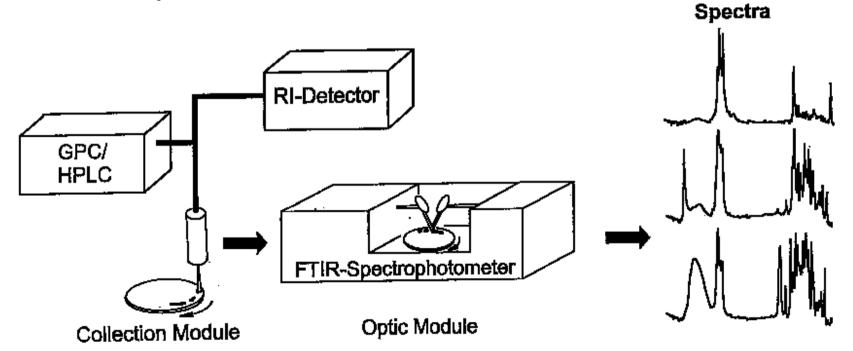




SEC with FTIR detection:

CCD, MMD information

Simultaneous separation and identification of fractions





SEC with FTIR detection:

CCD, MMD information

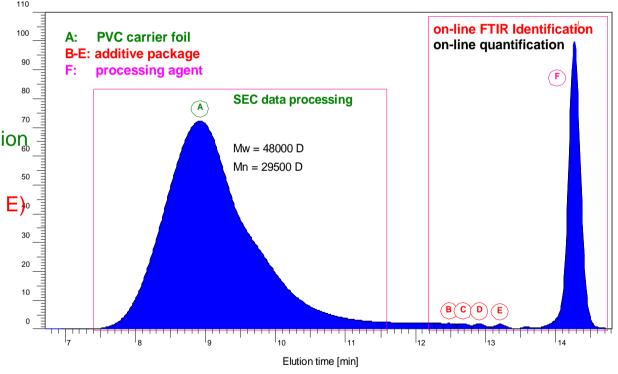
type and nature of the polymer used (peak A: PVC)

molar masses and molar mass distribution of the polymer (peak A)

identification of the additives (peaks B - E)

quantification of all additives in the packaging foil

identification and quantification of the processing agent (peak F)



SEC with MS Detection



Potential benefits

multiple MS techniques:

offline: MALDI

online: SQ, QQQ, QTOF, ion mobility

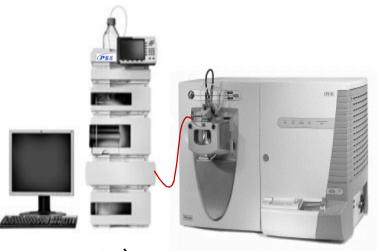
• MS measures M with highest accuracy

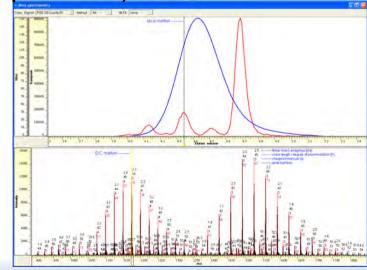
MS has high resolution

MS can resolve co-eluting species

MS offers very high sensitivity (trace components)

current instruments easy to use





SEC with MALDI-MS Detection



Overview

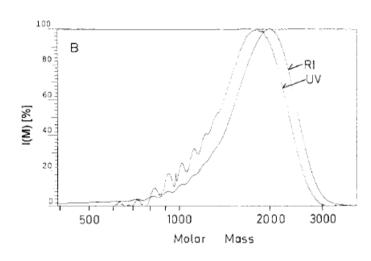
MALDI advantages:

- absolute molar mass
- repeat unit identification
- endgroup determination
- -structure elucidation
- high molar mass range

disadvantages:

- matrix influnces
- discrimination in polydisperse samples
- only offline mode (spotting)
- copolymers difficult

GPC-MALDI of PMMA Ref.: Gores, Pasch; Polymer **36**, 1999



Matrix: 10,21 mg DHB / 1 ml THE Kratos Kompact MALDI 3 V3.0 Run PMMA0038 4 Jan 94 21:17 +Ref Hi Pwr 61 Sample 16: 4,5 mg PMMA 2030 / 1 ml THE

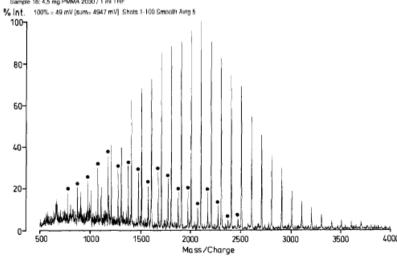


Figure 4 M.a.l.d.i.-m.s. spectrum of a PMMA calibration standard (sample 3); full circles indicate the cyclic oligomers

SEC with ESI-MS Detection



Overview

online MS advantages:

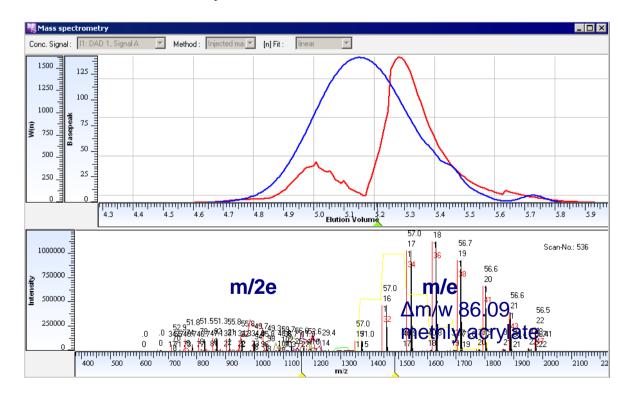
- absolute molar mass
- repeat unit identification
- endgroup determination
- structure elucidation
- -identification
- high resolution

disadvantages:

- hmw limitations
- multiple charges (ESI)
- copolymers difficult

Current state of SEC-MS:

- integration in GPC/SEC software
- easy-to-use for chromatographers
- many automated workflows

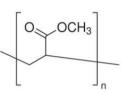


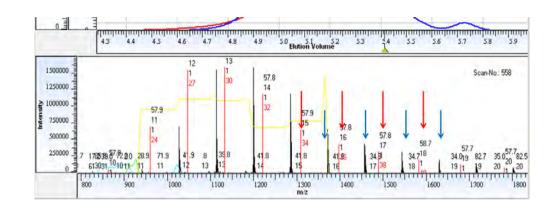
SEC with ESI-MS Detection



Analytical Results

- poly(methyl acrylate)
- 2 endgroups propyl (43), butyl (57)
- simple charge pattern
- good mass resolution
- 2 main distributions same MA repeat units

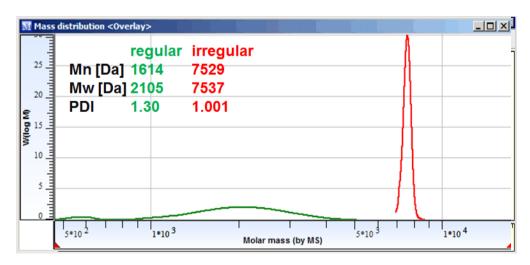




SEC-MS reveals:

- mixture of species which behave differently in SEC
- GPC separation of regular chains
- no separation if irregular species

However: still absolute M and MWD

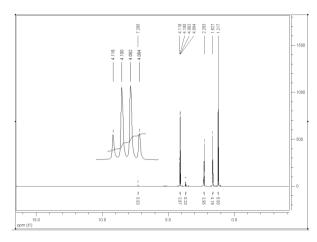


SEC with NMR Detection



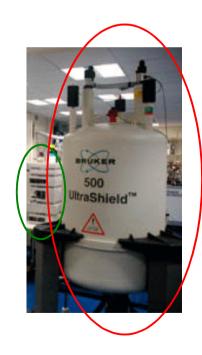
Basics

NMR can be used as a (universal) chemical detector NMR is a chemical sensor looking at local chemical environment ideal for structure elucidation: chemical shift, J coupling



High-field NMR couling

non-destructive
super-conductivemagnet
high resolution
small differences obvious
expensive
complex
large
time-consuming
interfacing difficult
high operational cost
expert knowledge required



Low-field NMR detection

non-destructive
permanentmagnet
lowresolution
major sample characteristics
inexpensive
simple to use (detector)
small benchtop
low operation cost
flexible
modular setups
saves sample prep time



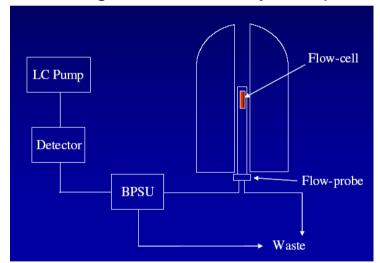
SEC with NMR Detection



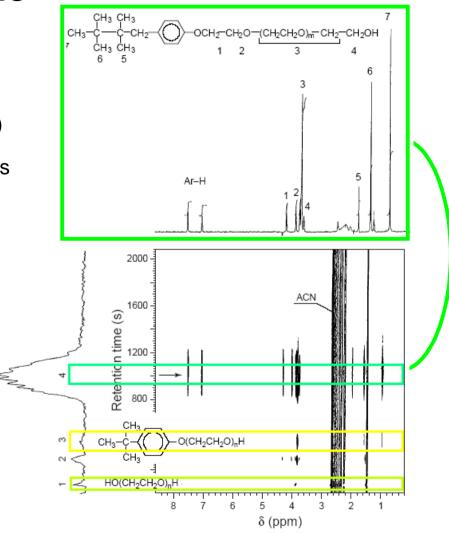
High-Field NMR Coupling to HPLC

Sample: non-ionic surfacrants

- samples have been stored in storage valve (BPSU)
 Offline NMR scans
- solvent signals eliminated by NMR pulse sequences



Ref.: Pasch/Hiller (1996), Macromolecules, 2, 6556



SEC with ¹H-NMR Detection



Current Status

base NMR: Bruker TopSpin, 20 MHz magnet

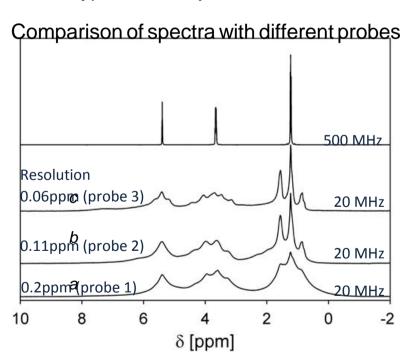
automatic supression of solvent peaks

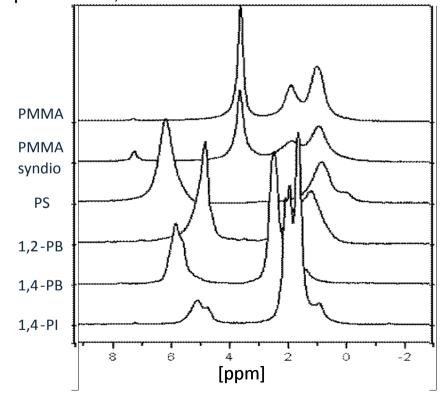
0.2ml probe

scan time: 2 secs

run on: PSS SECcurity GPC system, single PSS SDV 5µm column, THF

typical SEC injection conditions







Conclusions

- comprehensive SEC/GPC is an established and versatile method
- plethora of LC and detection methods for structure investigation
- -information request determines chromatographic strategy
- in-depth characterization of MMD, CCD, FTD, MAD, etc. possible
- combination of LC modes opens new horizons
- increase of peak capacity by 2D chromatography
- unbiased investigation of property distributions
- mapping of samples or property quantification in 2D
- information-rich detectors add identification and structure elucidation to separation